Synthesis of Dimeric $1,3,2\lambda^5$ -Benzoxazaphospholes from Phosphinic Acid Derivatives by Silylation/Desiloxylation

Alexander N. Bovin, Alla N. Yarkevich, and Eugene N. Tsvetkov*

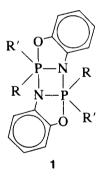
Institute of Physiologically Active Substances, USSR Academy of Sciences, Chernogolovka, 142432 Moscow region USSR

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ABSTRACT

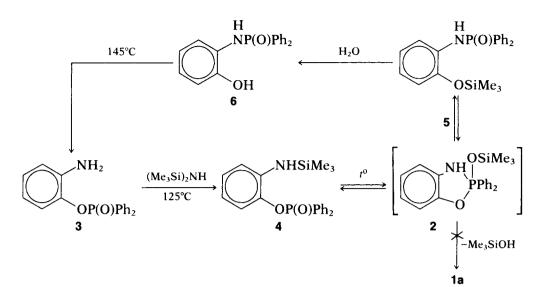
The reaction of N,N,O-tris(trimethylsilyl)-o-aminophenol with two equivalents of phosphinic chloride yielded dimeric 2,2-disubstituted 1,3,2 λ^5 -benzoxazaphospholes and trimethylsilyl phosphinate. The chlorides having the bulk substituents (o-chlorophenyl or tert-butyl) at phosphorus or containing P—N and P—O bonds (instead of a P—C bond) either didn't react at all or reacted to retain the phosphoryl group. Being stable in solution at 20°C, the individual diastereoisomers of dimeric 1,3,2-benzoxazaphospholes were converted upon warming to an equilibrium mixture of isomers. When reacted with another dimer each gave a mixed dimeric compound having two different phosphorus atoms in the molecule.

Dimeric 2,2-dialkyl(or diaryl)-1,3,2-benzoxazaphospholes (1) were synthesized by different routes; for example, by reaction of *o*-azidophenol with chlorophosphines [2], by dealkoxylation of 2-alkoxy-2,3-dihydro-2,2-diphenyl-1,3,2-benzoxazaphosphole (2) [3], or by reaction of *o*-aminophenol with trichloro- or trifluorophosphoranes followed by treatment with base [2, 4, 5]. The reaction of 2alkyl-2,3-dihydro-1,3,2-benzoxazaphosphole with acrylonitrile [6] or bis(diethylamino)methane [7] could serve as a more specific method for preparing the dimers 1 containing P—C bonds.



Phosphoryl compounds were not employed as precursors of dimers 1, though it is known that 1,3,2benzoxazaphosphole-containing phosphoranes, for example, may be prepared from o-aminophenol and phenyl dichlorophosphate [8]. On the other hand, the conversion of catechol-containing phosphates into the corresponding phosphoranes may be facilitated by silvlation of the oxygen on phosphorus [9]. Having taken these facts into account, we attempted to prepare the dimer 1a (R = Ph) via the hypothetical phosphorane 2. However, the silvlation of o-aminophenyl diphenylphosphinate 3 with subsequent heating to 200°C did not lead to dimer 1a. Instead, the reaction mixture contained only silyl derivatives 4 and 5 in a 4:5 ratio, as revealed by ³¹P NMR spectroscopy, from which amide 6 was isolated in 52% yield after hydrolysis. Thus, the result of the reaction is a reversible migration of the diphenylphosphinoyl group from oxygen to nitrogen. Interestingly, the isomerization of 6 into 3, for which severe conditions ($\sim 200^{\circ}$ C) are described by Cadogan [3], proceeded upon refluxing in xylene.

^{*}To whom correspondence should be addressed.



By silvlation of the supposed intermediate phosphorane 2 with trimethylchlorosilane under more severe conditions (autoclave $230^{\circ}C \times 6$ h), we prepared the dimer 1a in 41% yield (from ester 3):

4
$$\xrightarrow{\text{Me}_3\text{SiCl, NEt}_3}_{\text{autoclave, 230°C}}$$

-NEt₃ · HCl
 $\overrightarrow{\text{OSiMe}_3}$
 $\overrightarrow{\text{OSiMe}_3}$

Because of the experimental inconvenience of the autoclave method, we also prepared dimer **1a** and related ones (Table 1) by a simplified method. In this case, the mixture of N,N,O-tris(trimethylsilyl)o-aminophenol (7) and chloride **8** (1:2 ratio) was refluxed in xylene to give (after removal of Me₃SiCl) the dimers **1a-h** and trimethylsilyl phosphinate as a byproduct:

With an equimolar ratio of 7 and 8, the hydrolysis of the reaction mixture, after removal of dimer 1, gave its o-hydroxyphenylammonium salt as demonstrated for dimers 1g,h, not the phosphinic acid (from RR'PO₂SiMe₃). The compounds **1b-e** had already been formed at 110°C (under refluxing in toluene), whereas heating to 200°C was necessary to complete the conversion of 9a to 1a. The dimers 1d-g ($R \neq R'$) were obtained as a mixture of two diastereoisomers. In this case, the formation of the intermediate imidodiphosphinate 9 was confirmed by the observation of pairs of equal signals at $\delta^{31}P = +52.17, 52.29 \text{ and } +50.72, 50.84, \text{ correspond-}$ ing to diastereomers 9d,e in the ³¹P NMR spectra of reaction mixtures of 1d,e (formed after refluxing for 1-3 h in toluene).

At least for the dimers **1b–e**, the conversion of **9** into **1** became reversible at high temperature (>170°C) and possibly at high concentration. It was demonstrated that after heating (200°C, 5 h) the mixture of Ph₂PO₂SiMe₃ and **1b** or **1e** in nitrobenzene gave ³¹P NMR spectra with the signals at $\delta =$

Compd.	R, R'	Yield,* %	Mp, °C (solvent)	³¹ P{ ¹ H} ^b (nitrobenzene)	Mass Spectrum, m/z, M ⁺ and 1/2 M ⁺ (rel. intensity)	
1a	Ph ₂	69	226–227° dec.	-50.9 ^d	582(8), 291(100)	
1b	Et ₂	77	157–160° (toluene)	-33.8′	390(12), 195(100)	
1c	Octyl ₂	59	82–85 (hexane) ^g	-36.2 ^r	726(2), 363(50)	
1d	Me, Et	64	175–177 [/] (benzene)	-39.4, -38.7 ⁱ	362(7), 181(100)	
1e	Me, Bu	74	144–148 ^{//} (diethyl ether)	-40.5, -40.1 ⁱ	418(34), 209(100)	
1f	Me. Ph	73	165–185 [/]	-49.1, -47.9	458(4), 229(100)	
1g	o-CH₂OC ₆ H₄	33 ^k	230–250 [/] dec.	-36.9, -35.5 ¹	486(0), 243(60)	
1f	o-(CH₂O)₂C ₆ H₄	41 [*]	225–230 dec.	-42.4	546(8), 273(100)	

TABLE 1 Selected Data of Dimeric 1,3,2-Benzoxazaphospholes 1a-h

"Based on 7; for 1b-e after column chromatography.

^b Downfield signals (δ +20-45) of possible monomers in nitrobenzene are not indicated.

^d In all cases the signal at δ -41.6 also was registered.

° 169-170 [2].

' CDCl₃.

⁹ Recrystallized at -10°C.

^hLess soluble isomer.

¹ Minor (more soluble) and major (less soluble) isomers in C₆D₆.

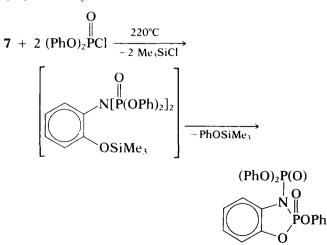
/ Mixture of isomers.

* At 1:1 ratio of 7 and 8.

¹ In all cases the signals at δ -45.3, and -45.0 also were registered.

+49 or +45 corresponding to $Et_2PO_2SiMe_3$ and $BuMePO_2SiMe_3$. In both cases, the insoluble precipitate of **1a** was also obtained.

The attempts to prepare dimers from bis(ochlorophenyl)- or di-*tert*-butylphosphinoyl chloride as well as tetraethylphosphorodiamidic chloride were unsuccessful. Under the same conditions, diphenyl chlorophosphate gave 2-oxo-2-phenoxy-3-diphenoxyphosphoryl-1,3,2-benzoxazaphosphole (**10**) in 90% yield:



10

The less soluble major diastereomers 1d,e were isolated from the mixture of isomers by crystallization. X-ray analysis of the isomer 1e [10] demonstrated the trans orientation of butyl (or methyl) substituents relative to the diazadiphosphetidine ring. Having compared the ³¹P NMR spectra of 1d,e, we assigned a trans configuration to the major isomer 1d.

Being stable in solution at 20°C, the isomers **1d,e** were converted into an equilibrium mixture (1:3) of cis- and trans-dimers upon warming in benzene (75°C, 2–3 h) as revealed by ³¹P NMR spectroscopy. In this connection, the recrystallization of the major isomer **1d** from benzene sometimes resulted in its partial isomerization. The reaction between two different dimers proceeded as easily as their isomerization. For example, the mixed dimer **1be** (from **1b** and **1e**) was obtained in benzene after heating (75°C, 3 h) and appeared in the ³¹P NMR spectrum as a pair of doublets (${}^{2}J_{pp}$ 18 Hz). The ³¹P NMR data of **1be** and other mixed dimers obtained in solution are listed in Table 2.

EXPERIMENTAL

NMR spectra (chemical shifts from internal Me₄Si for ¹H and from external H_3PO_4 for ³¹P{¹H}; positive for downfield shifts) were recorded on a Bruker CXP-200 spectrometer at 200 and 81 MHz, respectively.

^{° 210-220 [3]}

Compd.	R^1 , R^2 at P^1	R ³ , R ⁴ at P ²	δP^1	δP^2	²J _{₽₽} , Hz
1 be ^a	Et ₂	Me, Bu	-33.95	-40.28	18
1ag [⊳]	Ph ₂	o-CH₂OC ₆ H₄	-51.34	-37.98	29
1de ^a	Me, Et	Me, Bu	-40.56	-39.44	20
1gh°	0-(CH ₂ O) ₂ C ₆ H ₄	o-CH₂OC ₆ H₄	42.74	-37.31	29
1df⁵	Mè, Et	Me, Ph	-38.30	-48.90	20
^e C ₆ D ₆ . ^b Nitroben ^c Benzonit					

TABLE 2 ³¹P{¹H} NMR Spectra of Mixed Dimeric 1,3,2-benzoxazaphospholes

 Prepared by Heating of Mixture of Dimers 1a,b,d and 1e-h in Solution

Mass spectra were obtained on a Finnigan 4021 chromatomass spectrometer. All reactions involving silyl derivatives were carried out under argon. Chromatographic separations were performed on silica gel (eluent: benzene). All new compounds exhibited acceptable elemental analysis data.

o-Aminophenyl Diphenylphosphinate (3)

To a stirred solution of 5.7 g (52 mmol) of *o*-aminophenol and 5.3 g (52 mmol) of NEt₃ in 30 mL of dry dioxane, 12.4 g (52 mmol) of diphenylphosphinoyl chloride was added dropwise. The mixture then was heated for 30 min at 50°C. After removal of a precipitate, the solution was concentrated to dryness *in vacuo*, and the residue was crystallized from benzene to afford 13.1 g (81%) of **3**, mp 114–116°C; ³¹P NMR (CDCl₃), δ 32.9; ¹H NMR (CDCl₃), δ 4.49 (s, NH₂), 6.57, and 6.91 (m, 1 + 3H, C₆H₄), 7.50, and 7.94 (m, 6 + 4H, C₆H₅).

N-(2-Hydroxyphenyl)-diphenylphosphinic Amide (**6**)

A mixture of 1.3 g (4.2 mmol) of the ester **3**, 0.01 g of $(NH_4)_2SO_4$ as a catalyst, and 6 mL of $(Me_3Si)_2NH$ was refluxed until the solution became transparent (2 h). To the resulting solution, which gave a 31 P NMR spectrum with two signals: δ 30.6 and 15.7 (11:1 ratio), 10 drops of dry pyridine were added. The mixture then was heated for 1 h to 170-200°C with simultaneous distillation of solvent. The ³¹P NMR spectrum of the residue obtained showed a 4:5 ratio of the same signals (δ 34.0 and 19.0 in CDCl₃). The residue was dissolved in 15 mL of acetone and allowed to stand for 20 h. The resultant precipitate was collected and crystallized from 2butanone-ethanol to afford 0.7 g (52%) of 6, mp 224-227°C (228-231°C [3], 219-222°C [5]), ³¹P NMR (DMSO), δ 19.74; ¹H NMR (DMSO-D₆), δ 6.58, 6.80, and 7.04 (m, 1 + 2 + 1H, C₆H₄), 6.92 (d, ²J_{HP} 11 Hz, NH), 7.56, and 7.84 (m, 6 + 4H, C₆H₅), 9.86 (s, OH).

Conversion of Amide 6 into Ester 3

A mixture of 0.8 g (2.6 mmol) of amide 6 and 30 mL of xylene was refluxed for 15 h in a Dean-Stark ap-

paratus. The resultant solution was concentrated by rotary evaporation, and the residue was crystallized from ether to yield 0.7 g (84%) of ester **3**.

Preparation of Dimer **1a** from Ester **3** by Silylation in an Autoclave

A mixture of 5.5 g (18 mmol) of ester **3**, 0.01 g of $(NH_4)_2SO_4$, and 8 mL of $(Me_3Si)_2NH$ was refluxed for 2 h to form a transparent solution and was then concentrated *in vacuo*. The resultant oil was dissolved in 15 mL of toluene (or acetonitrile) and heated with 2.9 g (27 mmol) of Me_3SiCl and 3.6 g (35 mmol) of NEt₃ in a sealed autoclave tube (230°C, 6 h). After having been cooled to 20°C, the resultant solid was collected and washed with benzene (20 mL) and ethanol (3 × 20 mL) to afford 2.2 g (41%) of **1a**.

N,N,O-tris(Trimethylsilyl)-o-aminophenol (7)

A mixture of 38.0 g (350 mmol) of *o*-aminophenol, 0.05 g of $(NH_4)_2SO_4$ and 150 mL of $(Me_3Si)_2NH$ was refluxed for 10 h. The resulting solution was cooled to 20°C, and then 13.0 g (120 mmol) of Me_3SiCl and 12.0 g (120 mmol) of NEt_3 were added. After 2 days the mixture was filtered under argon and the filtrate was distilled to afford $88.0 \text{ g} (\sim 100\%)$ of *N*,*O*-bis(trimethylsilyl)-*o*-aminophenol, bp $120-121^{\circ}C$ (10 mm Hg).

To a stirred solution of 37.0 g (146 mmol) of N,O-bis(trimethylsilyl)-o-aminophenol in 20 mL of anhydrous ether was added dropwise to a solution of phenyllithium prepared from 2.4 g (350 mg-at) of Li and 30.6 g (195 mmol) of bromobenzene in 150 mL of ether. To the resultant mixture, which was cooled to 0°C, was added 21.2 g (195 mmol) of Me₃SiCl. This solution was stirred 2 h at 20°C and filtered. The filtrate was distilled to give 43.0 g (91%) of 7, bp 102–103°C (2 mm Hg), n_D²⁰ 1.4812. ¹H NMR spectrum matched that of Narkon [11].

$[1,3,2\lambda^5,4\lambda^5]$ Diazadiphospheto[2,1-b:4,3-b']bis[1,3,2]benzoxazaphospholes (**1a-h**): Typical Procedure

To a solution of 3 mmol of 7 in 5 mL of dry xylene was added 6 mmol of phosphinic chloride 8. This

mixture was refluxed for 4-5 h or, in the case of 8a, heated to 200°C with simultaneous distillation of volatile products. Then the mixture was cooled to 20°C, diluted with dry benzene (10 mL), and either filtered to yield the precipitate of 1a,f-h (after washing with benzene and dry DMSO) or chromatographed to afford 1b-e. At a 1:1 ratio of 7 and 8g,h, the corresponding o-hydroxyphenylammonium phosphinates were isolated from the mother liquor obtained after hydrolysis: 3-hydroxy-3-oxo-2H-1,3-benzoxaphosphole, o-hydroxyphenylammonium salt (C₇H₇O₃P·C₆H₇NO, mp 206–210°C dec. (wet ethanol)) and 3-hydroxy-3-oxo-2H, 4H-1,3,5benzodioxaphosphepin. o-hydroxyphenylammonium salt (C₈H₉O₄P·C₆H₇NO, mp 175°C (dioxaneethanol)). At a 1:2 ratio of 7 and 8a, diphenylphosphinic acid (mp 192-195°C) was isolated in 58% vield after removal of dimer la and subsequent hydrolysis. For 1b ¹H NMR (CDCl₃), δ 1.22 (dt, ³J_{HH} 7 Hz, ³J_{HP} 22 Hz, 12H, CH₃), 2.25 (quintet, 8H, CH₂), 6.60 and 6.72 (m, 2 + 6H, C_6H_4). For 1c ¹H NMR $(CDCl_3)$, $\delta 0.87$ (t, 12H, CH₃), 1.25 (m, 4OH, $(CH_2)_5$ Me), 1.63 (m, 8H, PC---CH₂), 2.19 (m, 8H, PCH₂), 6.58 and 6.73 (m, 2 + 6H, C₆H₄). For 1d (trans) ¹H NMR (C_6D_6) , δ 0.85 (dt, ${}^{3}J_{HH}$ 7 Hz, ${}^{3}J_{HP}$ 24 Hz, 6H, C--CH₃), 1.62 (d, ${}^{2}J_{HP}$ 13 Hz, 6H, PCH₃), 1.94 (m, 4H, PCH₂--C), 6.37, 6.78, and 6.94 (m, 2 + 4 + 2H, C_6H_4). For 1e (trans) ¹H NMR (CDCl₃), δ 0.85 (t, 6H, $C-CH_3$, 1.32 and 1.55 (m, 4H + 4H, $C-(CH_2)_2-C$), 1.99 (d, ²J_{HP} 13 Hz, 6H, PCH₃), 2.23 (m, 4H, PCH_2 —C), 6.64 and 6.72 (m, 2 + 6H, C₆H₄). Mp's, ³¹P NMR and mass spectra of **1a-h** are in Table 1.

2-Oxo-2-phenoxy-3-diphenoxyphosphoryl-1,3,2benzoxazaphosphole (10)

A solution of 4.4 g (16 mmol) of (PhO)₂P(O)Cl and 2.7 g (8 mmol) of **7** in 10 mL of dry xylene was heated for 5 h at 220°C with simultaneous distillation of volatile products. Then the reaction mixture was concentrated *in vacuo* at 150°C to yield 3.5 g (90%) of **10** as a viscous oil, which was induced to crystallize from ether. Recrystallization from toluene gave the compound **10** as a very hygroscopic solid, mp 98–100°C, ³¹P NMR (CDCl₃), δ –14.6 d, +6.8 d (²J_{PP} 24.4 Hz); ¹H NMR (CDCl₃), δ 7.25 (m, 14H), 7.34 (m, 4H), 7.45 (m, 1H).

Reaction of **1b**,**e** with Trimethylsilyl Diphenylphosphinate: Typical Procedure

A solution of 0.6 mmol of trimethylsilyl diphenylphosphinate (prepared by refluxing of Ph₂PO₂H in (Me₃Si)₂NH) and 0.6 mmol of **1b**,**e** in 1.5 mL of nitrobenzene was heated in a sealed NMR tube for 5 h at 200°C. The substantial signals of Et₂PO₂SiMe₃ (δ + 49.0) or BuMePO₂SiMe₃ (δ + 45.0) were detected and matched the ³¹P NMR spectra of separately prepared silyl esters. In both cases the insoluble precipitate of **1a** was formed.

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